Soil Fertility Update

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## Phosphorus Math for Dummies

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## Comparing the costs of $\mathrm{P}_{2} \mathrm{O}_{5}$ in different phosphorus fertilizers?

Several reasons come to mind why we would want to calculate the value of $\mathrm{P}_{2} \mathrm{O}_{5}$ in different phosphorus $(P)$ fertilizers. One question we are asked frequently concerns the cost of 10-34-0 when used as a starter fertilizer. Can the same starter benefit be achieved with liquid $N$ without 10-34-0? If not, what extra cost is incurred with 10-34-0? Another common question is which is a cheaper source of $P$; monoammonium phosphorus (MAP) or diammonium phosphate (DAP)? Maybe surprisingly, you cannot tell just from the cost per ton.

Back to the starter fertilizer question. Our recent research has shown $2 \times 2$ starter fertilizer results in lower harvest grain moisture in most situations and increased yield in some cases. ${ }^{1}$ Most of our trials compared mixes of $2 \times 2$ placed urea ammonium nitrate (UAN) and $10-34-0$ to a no starter control. Quite a lot of earlier research found a synergy between coapplied nitrogen $(N)$ and $P$ that stimulated plant growth and increased $N$ and $P$ uptake. In our work, we did not try to distinguish whether the $P$ component was necessary, but there is good reason to think the combination is better than N alone. In any case, some growers want to leave it out because it increases the cost of the starter fertilizer. How much does the $\mathrm{P}_{2} \mathrm{O}_{5}$ in 10-34-0 really cost compared to $\mathrm{P}_{2} \mathrm{O}_{5}$ in typical broadcast P fertilizers - MAP, DAP or triple super phosphate (TSP)?

In the following text we outline the approach to figuring the comparative cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ from the different fertilizers and the accompanying Excel spreadsheet allows easy calculation of these values. Briefly, in any comparison one needs to subtract out the value of the N from the per ton cost of the fertilizer, then calculate the $\mathrm{P}_{2} \mathrm{O}_{5}$ cost. The N value is based on how much of the $N$ applied with the $P$ fertilizer is estimated to be available to the crop and the cost of N in an at-plant or sidedressed N application. Unfortunately, in most situations less than half of the N applied in late-fall, winter, or early-spring remains in the soil for crops planted in April and May so it should not always be fully credited. After crediting the value of N in the P fertilizer, the remaining cost of the fertilizer is attributed to the $\mathrm{P}_{2} \mathrm{O}_{5}$ content.

If you don't want to learn how to do the calculations by hand, skip to the end of the article and download the Excel spreadsheet.
${ }^{1}$ Summary of Starter Fertilizer Trials - 2014-2018.
https://ag.purdue.edu/agry/extension/Documents/Soil\ Fertility/Phosphorus\ Math\ for\ Dummies.pdf

## Comparing $\mathrm{P}_{2} \mathrm{O}_{5}$ cost from 10-34-0 and DAP

Written out below are examples comparing the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ from 10-34-0 to the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ from DAP, crediting $0 \%, 50 \%$, or $100 \%$ of the DAP $N$ value.

## Step 1: Calculate the cost of a pound of $\mathbf{N}$ from UAN.

In this example, UAN $28 \%$ costs $\$ 285$ per ton and contains 560 pounds of N . Therefore the price per pound of N is $\$ 0.51$ ( $\$ 285$ per pound divided by 560 pounds).

## Step 2: Subtract the value of $\mathbf{N}$ in 10-34-0 from the per ton price, based on the price of UAN.

In this example, 10-34-0 costs $\$ 495$ per ton and contains 200 pounds of $N$ and 680 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$.

200 pounds of $N$ times $\$ 0.51$ per pound equals $\$ 102$. Thus $\$ 393(\$ 495-\$ 102)$ of the cost of a ton of 10-34-0 is attributed to the $\mathrm{P}_{2} \mathrm{O}_{5}$ content of the 10-34-0.

Step 3: Calculate the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ in 10-34-0 after subtracting out the $\mathbf{N}$ value.
$\$ 393$ per ton divided by 680 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$ per ton equals $\$ 0.58$ per pound of $\mathrm{P}_{2} \mathrm{O}_{5}$.

## Calculating $\mathrm{P}_{2} \mathrm{O}_{5}$ cost from DAP with different credits for $\mathbf{N}$

We take the same approach with MAP or DAP, except the value of the $\mathrm{P}_{2} \mathrm{O}_{5}$ is also dependent on how much of the N is likely to be available to the crop. If the fertilizer is spread shortly before planting then most of the N would be available to the crop ( $100 \%$ credit), but if it is spread in late-fall or early-winter, then nearly all of the N could disappear before the crop needs it ( $0 \% \mathrm{~N}$ credit).

For example consider DAP (18-46-0) at $\$ 490$ per ton. DAP contains 360 pounds of N per ton and 920 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$ per ton.

## Calculating $\mathrm{P}_{2} \mathrm{O}_{5}$ cost from DAP with $\mathbf{0 \%} \mathbf{N}$ credit:

If $0 \%$ of the N is credited from a fall application of DAP, then calculating the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ in MAP or DAP is easy - just divide the cost of the fertilizer by its $\mathrm{P}_{2} \mathrm{O}_{5}$ content (identical to Step 1 above, except based on the $\mathrm{P}_{2} \mathrm{O}_{5}$ content and per ton price, only).
$\$ 490$ per ton divided by 920 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$ per ton equals $\$ 0.53$ per pound of $\mathrm{P}_{2} \mathrm{O}_{5}$.

## Calculating $\mathrm{P}_{2} \mathrm{O}_{5}$ cost from DAP with $\mathbf{1 0 0 \%}$ N credit:

If none of the N is lost from DAP the calculation is identical to that for 10-34-0 above (Step 13). First subtract out the value of N from the per ton cost and then assign the remaining cost to the $\mathrm{P}_{2} \mathrm{O}_{5}$ content of the fertilizer. With these prices, and no N loss, the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ from DAP is $\$ 0.33$ per pound.

## Calculating $\mathrm{P}_{\mathbf{2}} \mathrm{O}_{\mathbf{5}}$ cost from DAP with $\mathbf{5 0 \%} \mathbf{N}$ credit:

Research in Illinois and Minnesota has shown that we might expect $50 \% \mathrm{~N}$ loss of late-fall applied DAP or MAP with 'normal' winter weather. If we credit $50 \%$ of the $N$ to the next crop, then we deduct only $50 \%$ of the value of N from the cost of the fertilizer and then calculate the $\mathrm{P}_{2} \mathrm{O}_{5}$ cost as described in Step 1-3. DAP costs $\$ 490$ per ton and contains 360 pounds of N and 920 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$.

Step 1: If $50 \%$ of the N is assumed to be lost after application, we only credit 180 pounds per ton.

180 pounds of $N$ times $\$ 0.51$ per pound of $N$ from UAN equals $\$ 92$. Thus $\$ 398(\$ 490-\$ 92)$ of the cost of a ton of DAP is attributed to the $\mathrm{P}_{2} \mathrm{O}_{5}$ content of the DAP ( 960 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$ ).

Step 2: Calculate the cost of $\mathrm{P}_{2} \mathrm{O}_{5}$ in DAP after subtracting out $50 \%$ of the N value.
$\$ 398$ per ton divided by 920 pounds of $\mathrm{P}_{2} \mathrm{O}_{5}$ per ton equals $\$ 0.43$ per pound of $\mathrm{P}_{2} \mathrm{O}_{5}$.

Darn math! Instead of paper, pencil and a calculator, download this Excel spreadsheet, choose your N and $P$ sources, enter per ton prices, choose a level of N loss dependent on the timing of application and perceived N credit, and compare the per pound costs of $\mathrm{P}_{2} \mathrm{O}_{5}$ without strain on the brain. To get the comparisons on a per acre basis choose the comparison of interest and enter the rate of fertilizer to be applied. If you are specifically interested in the comparison of 10-34-0 to another $P$ source, then


N credit for $\mathbf{1 0 - 3 4 - 0}$ is typically $100 \%$ when applied as a starter fertilizer so the calculation of value is insensitive to N credit chosen for this source.
N credits should be chosen based on timing of application and weather. $100 \%$ if applied shortly before planting. $50 \%$ or $0 \%$ if applied in the fall or winter. Assumes difference in N application will be made up with the primary N source. only include the rate of 10-34-0 in gallons to determine the cost differential per acre. Any questions or problems please email jcambera@purdue.edu.

AGRONOMY

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