



Nutrient Management for Vegetable Crops

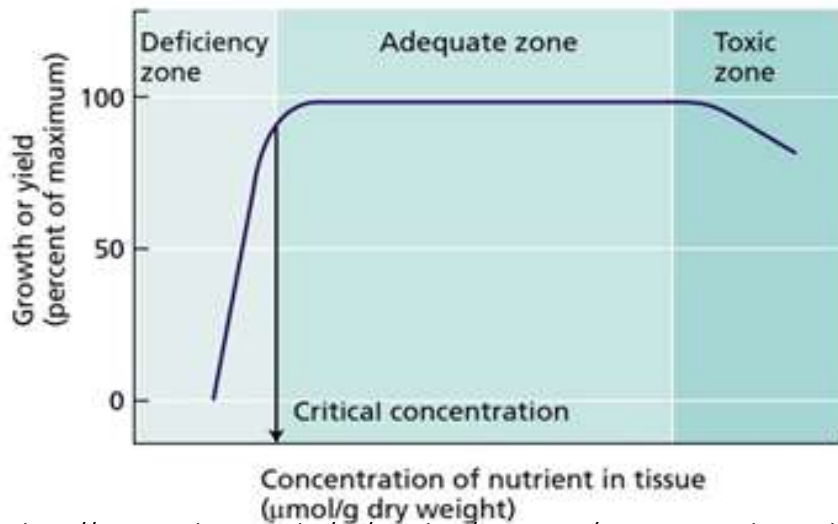
Plant Nutrients and Fertilizer Sources

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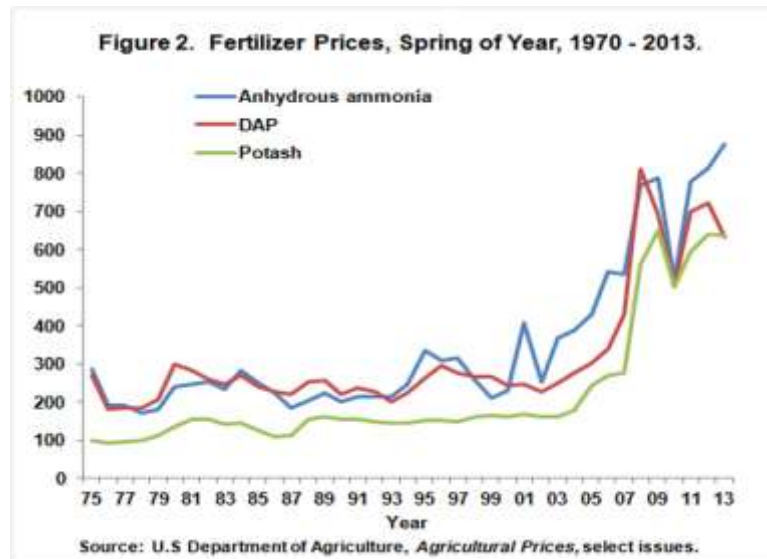
January 11, 2017

❖ Goals of Plant Nutrient Management Plan

- ✓ Ensure supply of essential nutrients dose not limit crop growth
- ✓ Optimizing the economic returns from fertilizers
- ✓ Minimize the negative impact of nutrients on the environment



http://www.tankonyvtar.hu/en/tartalom/tamop425/0010_1A_Book_angol_01_novenyelettan/ch02s02.html



Source: U.S Department of Agriculture, *Agricultural Prices*, select issues.
<http://farmdocdaily.illinois.edu/2014/01/controlling-costs-with-lower-crop-revenues-fertilizer-costs.html>



<http://extension.usu.edu/aces/htm/fertilizer-management>

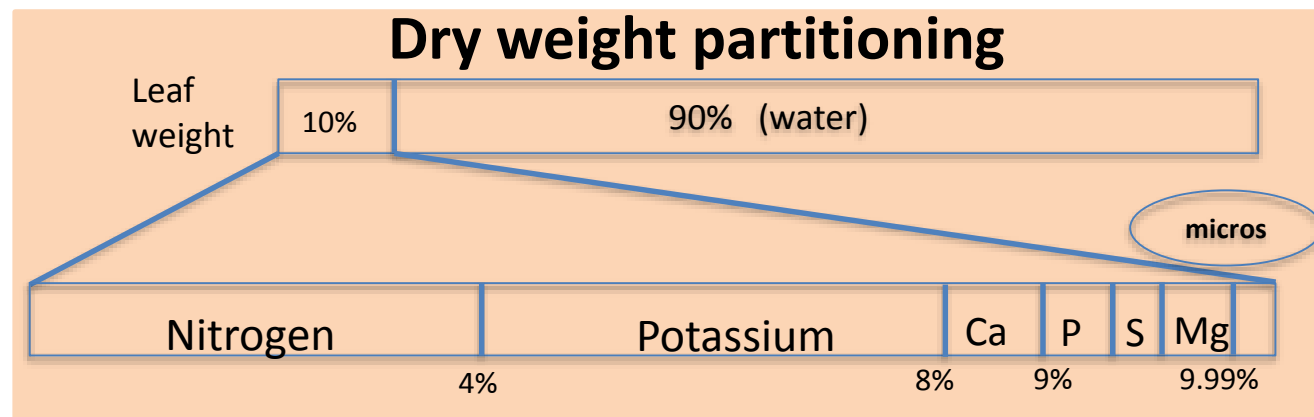
❖ Essential Nutrients

Macronutrients:

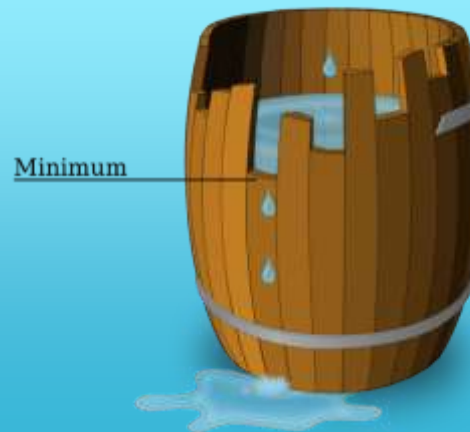
- Nitrogen N
- Phosphorus P
- Potassium K
- Calcium Ca
- Magnesium Mg
- Sulfur S

Micronutrients:

- Boron B
- Chlorine Cl
- Copper Cu
- Iron Fe
- Manganese Mn
- Molybdenum Mo
- Zinc Zn



Which nutrient is the most important one?



A crop's yield is restricted by the lack of a single element, even though there may be sufficient quantities of all other essential nutrients

❖ Essential Nutrients

Macronutrients:

- **Nitrogen** **N**
- **Phosphorus** **P**
- **Potassium** **K**
- **Calcium** **Ca**
- **Magnesium** **Mg**
- **Sulfur** **S**

Micronutrients:

- **Boron** **B**
- **Copper** **Cu**
- **Manganese** **Mn**
- **Zinc** **Zn**
- **Chlorine** **Cl**
- **Iron** **Fe**
- **Molybdenum** **Mo**

❖ Nitrogen

✓ Nitrate (NO_3^-) and Ammonium (NH_4^+)

- *Plants take up N in both forms. Excess ammonium can cause ammonium toxicity*
- *Fertilizers can provide N in both forms. Nitrate is the main source of nitrogen under soil temperature above 55 °F and aerated conditions regardless of fertilizer use*

✓ Nitrogen lost

- *Ammonium, volatilization if not incorporated*
- *Nitrate: leach, denitrification*

✓ **Nitrogen Fertilizers**

- Anhydrous ammonia (82% N) **ammonium-N**
- Urea (46% N) **ammonium-N**
- Nitrogen solutions (UAN, 28-32% N)
Ammonium : nitrate = 3:1
- Ammonium sulfate (21% N), **ammonium,**
recommended to lower soil pH
- Ammonium phosphate (10-21% N), **ammonium,**
supply P, also contribute N
- Calcium nitrate (15.5% N), **nitrate**
- Potassium nitrate (14% N), **nitrate**



Fertilizer forms	Advantages	Disadvantages
Nitrate-N fertilizers	<ul style="list-style-type: none"> • Ready source of N when soil temperature is low, and soil nitrate is low • Supply important cations: K, Ca 	<ul style="list-style-type: none"> • More expensive • Subject to leach • Increase soil pH
Ammonium-N fertilizers	<ul style="list-style-type: none"> • More economic source of N fertilizers • Less leach concerns 	<ul style="list-style-type: none"> • Subject to volatilization • Under suboptimal growing conditions (low soil temperature, saturated, low soil pH) and excess application, plant uptake excessive ammonium may lead to ammonium toxicity • Excess uptake of ammonium may reduce uptake of K, Ca and Mg. • Soil acidification

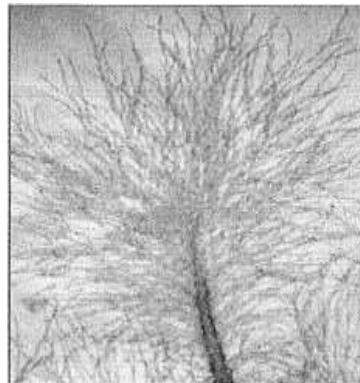
Nitrate form of nitrogen is considered when immediate source of nitrogen is needed while soil is cool (less than 55°F) and has low pH, or after water-logged soil has drained

❖ Phosphorus

✓ Most soil P is not available

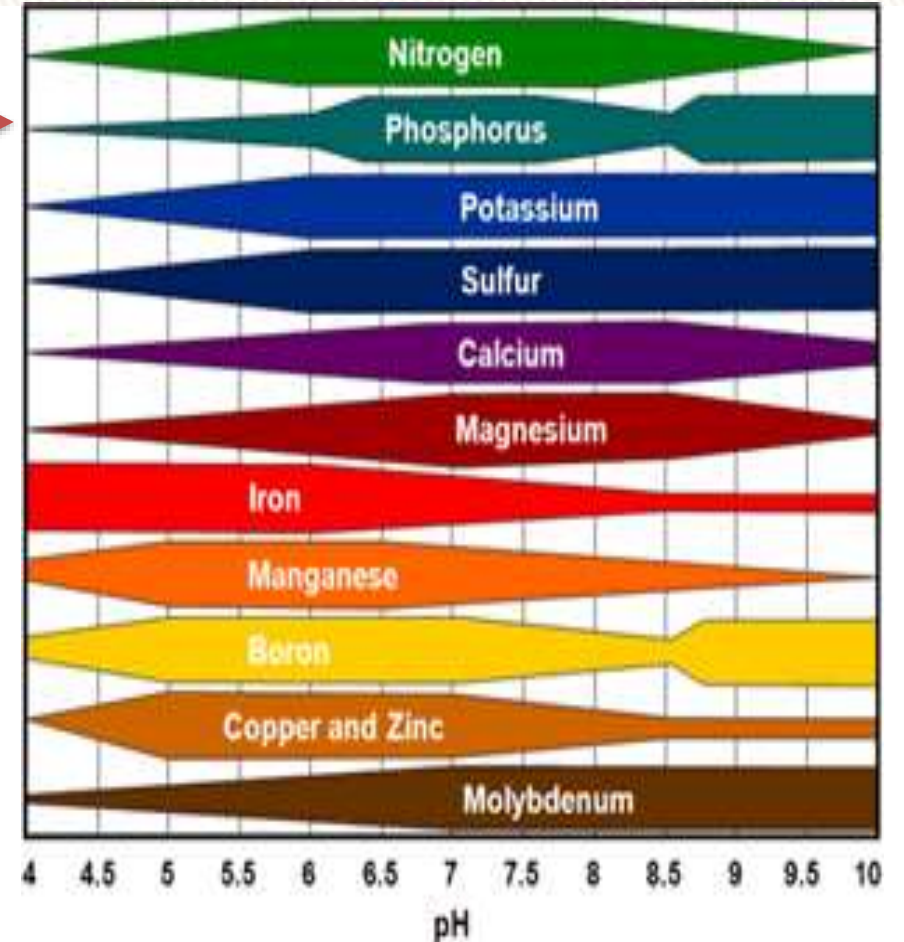
- Soil pH management
- Mobility of phosphorus is very limited.

Mycorrhizal fungi increase phosphorus uptake



<http://www.soillifeforyoursoilandcrop.com/about-mycorrhizal-fungi.html>

Purdue University is an equal access/equal opportunity institution.



<https://www.pioneer.com/home/site/us/agronomy/library/soil-sampling-test-interpretation/>

✓ **Apply P fertilizers**

- P is mostly applied in preplant fertilizer
 - Difficult to recover from phosphorus deficiency
 - Young seedlings are the most sensitive
 - Leaching is less a concern
- Band or broadcast
 - Soil with high phosphorus, broadcast is adequate

✓ Phosphorus fertilizers

- Monoammonium phosphate (MAP) (11% N, 48% P₂O₅)
- Diammonium phosphate (DAP) (18% N, 46% P₂O₅)
- Ammonium polyphosphate (APP) (10% N, 34% P₂O₅)

Phosphorus is more readily available when applied with nitrogen

- Superphosphate (20% P₂O₅)
- Rock Phosphate

❖ Potassium

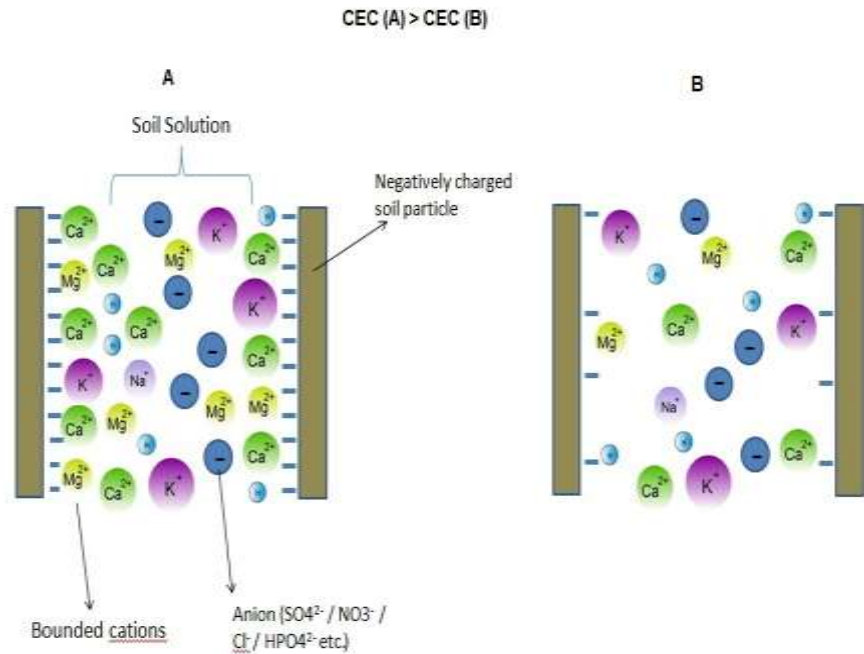
- Potassium is between N and P in mobility
- Leached in sandy soil, split application
- Recommendations takes into account Cation Exchange Capacity



✓ Potassium fertilizers

- Potash, potassium chloride (60-62 % K_2O)
High salt index, avoided if salinity is a problem
split apply if high rate is required
- Potassium sulfate (52% K_2O , 18% S)
- Sulfate of potash-Magnesia (K-Mag)(22% K_2O , 11% Mg, 22% S)
- Potassium nitrate (44% K_2O , 13% N)

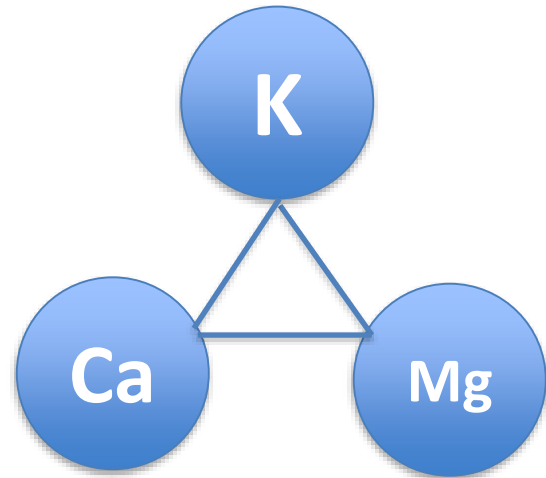
❖ Calcium and Magnesium



- Mg deficiency likely to occur in acid sandy soil with low CEC
- Most soil contain sufficient available calcium, soil moisture play a major role for plant to take up of calcium

<http://www.smart-fertilizer.com/articles/Cation-Exchange-Capacity>

✓ Antagonistic effects of K, Ca and Mg



- Excessive Ca may induce Mg deficiency
- A very high soil K can cause Ca and Mg deficiency
- High Mg and Ca may result in K deficiency

Base saturation: Ca at least 60%, Mg 10-15%, K 1-5%

Source: MSU E2934

✓ Calcium and Magnesium fertilizer sources

- Calcitic and dolomitic limes (3-12% Mg)
- *If pH is adequate*, apply Calcium sulfate (gypsum), and Magnesium sulfate (Epsom salt)
- Calcium nitrate (15.5% N, 19% Ca)
- Calcium chloride (36% Ca)
- Potassium magnesium sulfate (22% K₂O, 11% Mg, 22% S)

❖ Sulfur

- Primary source: soil organic matter, atmospheric deposition
- Sandy soil low in organic matter may show sulfur deficiency

✓ Sulfur fertilizer sources

- Potassium sulfate
- Gypsum
- Magnesium sulfate (Epsom salt)
- Element sulfur
- Ammonium sulfate

❖ Micronutrients

✓ Boron

Table 1. Potential for a crop response to boron when applied to boron deficient soils.

RESPONSE TO BORON		
LARGE	MODERATE	SMALL
Alfalfa, apple, broccoli, canola, cauliflower, celery, red beet, sugar beet, sunflower, turnip	Cabbage, carrot, clover, grape, lettuce, onion, radish, spinach, strawberry, and tomato	Asparagus, barley, blueberries, field corn, cucumber, oats, pasture grasses, pea, pepper, potato, raspberry, rye, soybean, sweet corn, and wheat

- Recommendations based on crops.
- Boron should be broadcast with other fertilizers
- Foliar application

✓ Boron fertilizer sources

- Borax
- Sodium tetraborate
- Boric acid
- Solubor
- Sodium pentaborate



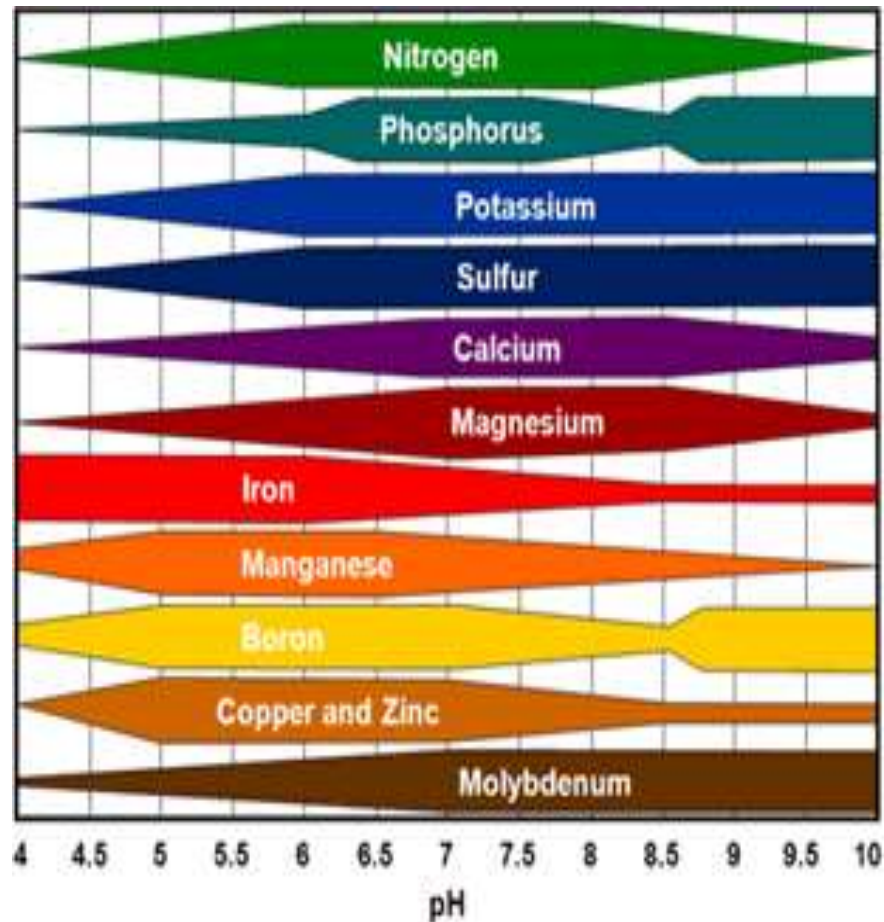
Figure 1. Areas of manganese deficiency.

Source: Purdue AY-276-W

- ✓ Some fungicide application supply Mn, Zn, Cu
- ✓ Manganese deficiency is likely on higher pH soils with higher levels of organic matter
- ✓ Zinc deficiency likely on higher pH soils with high P

Element	Source	Highly responsive crops
Mn	Mn sulfate	Bean, cuke, garlic, greens, lettuce, onion, pea, potato, pumpkin, radish, spinach, squash, sweet corn
Zn	Zn sulfate	Bean, onion, spinach, sweet corn

Source: MSU, e2934



❖ pH influence plant nutrient availability

- Optimal range pH 6.0-6.8
- Lime increase soil pH
 - Reduce Ca, Mg deficiency,
 - Reduce Mn toxicity
- Add element S or ammonium fertilizers reduce soil pH
 - Increase availability of Fe, Zn, Mn

<https://www.pioneer.com/home/site/us/agronomy/library/soil-sampling-test-interpretation/>

❖ Foliar fertilization

- Can be a viable way to supply a micronutrient if it is in unavailable form in the soil or soil application is not practical
- Not a practical approach for routinely supplying macronutrients

Possibly an option for overcoming temporary macronutrient deficiencies when soil conditions prevent root uptake (e.g. flooded soils)

❖ Fertilizer solubility

- When dissolve a fertilizer, not exceed its solubility

Solubility g/L	Fertilizer /Temperature (°F)	41	50	68	77	86	104
	Potassium nitrate	133	170	209	316	370	458
	Calcium nitrate	1020	1130	1290			
	Mono ammonium phosphate	250	295	374	410	464	567

- Fertilizer compatibility

Fertilizers contain calcium are incompatible with fertilizers contain phosphate and sulfate in a fertilizer concentrate

Calcium nitrate × Epsom salt

Calcium nitrate × Ammonium Sulfate

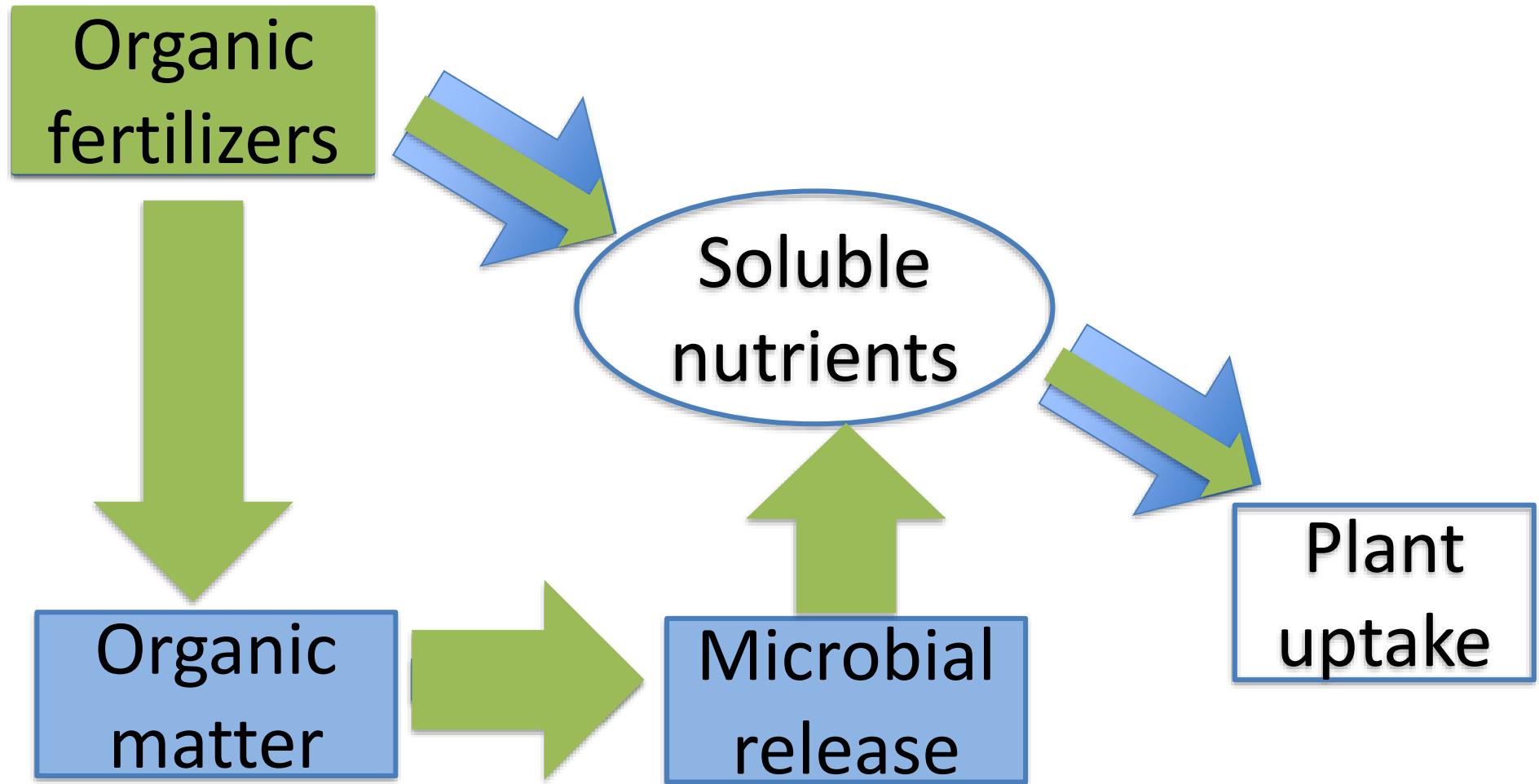
Calcium nitrate × Ammonium phosphate

Jar Test



<http://www.smart-fertilizer.com/articles/fertilizer-solubility>

❖ Plant nutrient management in organic systems



✓ **Manure and compost**

- Availability in current season:
 - 80% availability for P_2O_5
 - 90% availability for K_2O
 - Soluble Nitrogen plus 10-50% of N in organic form
- Manure should be applied at least 4 month prior to harvest
- Caution of building up P,

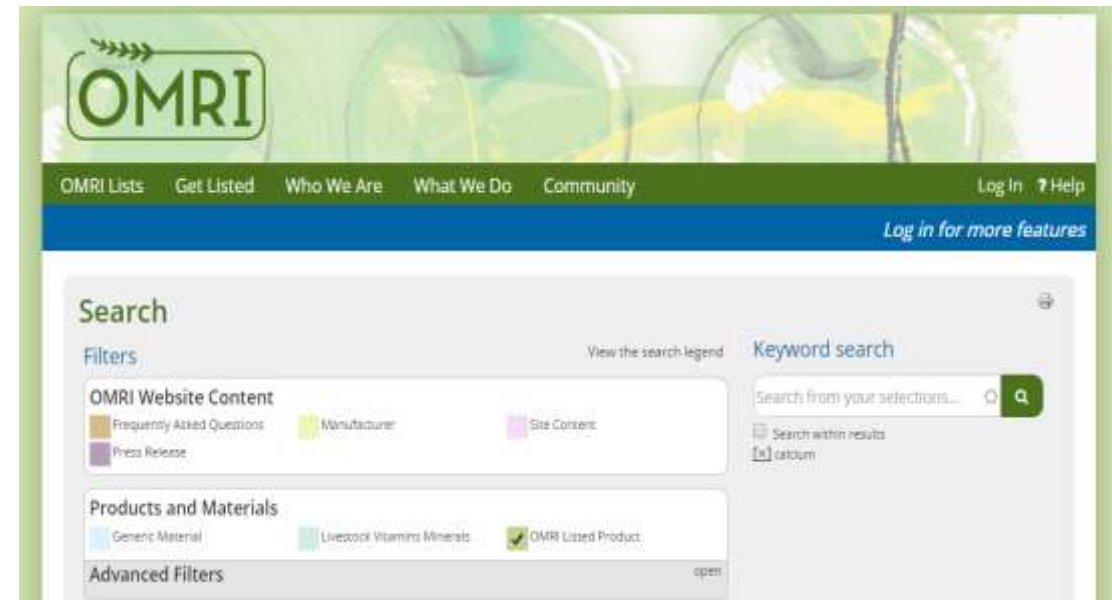
Compost: 1-3%N, 1-2%P, 1-2%K

✓ **Cover crops**

❖ Other fertilizer sources used in organic production

N	P	K
Feather meal	Bone meal	Greensand
Blood meal	Rock phosphate	Kelp meal
Fish product	Bat Guano	Potassium sulfate
Amino acid		Potassium magnesium sulfate

<http://www.omri.org/>



Questions & Comments

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